

Structure–Function Relationships and Crystal Structures of the Vitamin D Receptor Bound 2 α -Methyl-(20*S*,23*S*)- and 2 α -Methyl-(20*S*,23*R*)-epoxymethano-1 α ,25-dihydroxyvitamin D₃[†]Pierre Antony,[‡] Rita Sigüeiro,[§] Tiphaine Huet,[‡] Yoshiteru Sato,[‡] Nick Ramalanjaona,[‡] Luis Cezar Rodrigues,[§] Antonio Mouriño,[§] Dino Moras,^{*,‡} and Natacha Rochel^{*,‡}[‡]*Département de Biologie et de Génomique Structurales, Centre National de la Recherche Scientifique, Institut National de la Santé de la Recherche Médicale, Université de Strasbourg, CEBGS-IGBMC (Centre Européen de Biologie et Génomique Structurale—Institut de Génétique et de Biologie Moléculaire et Cellulaire), 1 Rue Laurent Fries, 67404 Illkirch, France and* [§]*Departamento de Química Orgánica and Unidad Asociada al CSIC, Universidad de Santiago de Compostela, 15782, Spain*

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The vitamin D nuclear receptor is a ligand-dependent transcription factor that controls multiple biological responses such as cell proliferation, immune responses, and bone mineralization. Numerous 1 α ,25(OH)₂D₃ analogues, which exhibit low calcemic side effects and/or antitumoral properties, have been synthesized. We recently showed that the synthetic analogue (20*S*,23*S*)-epoxymethano-1 α ,25-dihydroxyvitamin D₃ (**2a**) acts as a 1 α ,25(OH)₂D₃ superagonist and exhibits both antiproliferative and prodifferentiating properties in vitro. Using this information and on the basis of the crystal structures of human VDR ligand binding domain (hVDR LBD) bound to 1 α ,25(OH)₂D₃, 2 α -methyl-1 α ,25(OH)₂D₃, or **2a**, we designed a novel analogue, 2 α -methyl-(20*S*,23*S*)-epoxymethano-1 α ,25-dihydroxyvitamin D₃ (**4a**), in order to increase its transactivation potency. Here, we solved the crystal structures of the hVDR LBD in complex with the **4a** (C23*S*) and its epimer **4b** (C23*R*) and determined their correlation with specific biological outcomes.

Introduction

The pleiotropic actions of the hormonally active metabolite of vitamin D₃, the seco-steroid 1 α ,25-dihydroxyvitamin D₃ (**1**, 1 α ,25(OH)₂D₃, Figure 1), are mediated through its binding to the cognate nuclear vitamin D receptor (VDR^a), a ligand-dependent transcriptional modulator that belongs to the superfamily of steroid/thyroid hormone/retinoid nuclear receptors (NR).^{1–3} Among the structural multidomains of NR, the ligand binding domain (LBD), located at the carboxy terminus, is a key transcriptional regulator because it harbors the ligand induced transactivation function 2 (AF-2).⁴ Upon ligand complexation, the LBD undergoes a major conformational change involving the folding back of H12 onto the core of LBD, termed the “mouse-trap” mechanism.^{5,6}

Activated VDR controls multiple long-term biological responses including cell growth, apoptosis, angiogenesis, antiproliferation, differentiation, bone mineralization, and calcium/phosphate homeostasis.^{7–10} Therapeutic applications of pharmacological doses of 1 α ,25(OH)₂D₃, which encompass treatments for renal osteodystrophy, osteoporosis, psoriasis,

cancer, autoimmune diseases, and prevention of graft rejection are limited because of the hormone’s intrinsic hypercalcemic effect.¹¹ Therefore, the design of 1 α ,25(OH)₂D₃ analogues exhibiting antiproliferative and/or immunoregulatory properties with concomitant low calcemic side effect is the topic of intensive investigations at both mechanistic and clinical levels.^{12–16} The development of such analogues has led to more than 3000 compounds with structural modifications on the A and/or CD rings or the aliphatic side chain. However, only a few of these are used to treat human diseases.^{17,18}

In order to optimize the aliphatic side chain conformation with a subsequent entropy benefit, we recently designed and synthesized the 1 α ,25(OH)₂D₃ analogue, (20*S*,23*S*)-epoxymethano-1 α ,25-dihydroxyvitamin D₃ (AMCR277A, **2a**, C23*S*, Figure 1), which incorporates an oxolane ring in the side chain.¹⁹ We showed that it acts as a 1 α ,25(OH)₂D₃ superagonist because it is able to mediate transcriptional activity with a magnitude at least 10-fold higher than that of the natural ligand. In addition, it has the capacity to inhibit the proliferation of the human promyelocytic HL60 cells and induce their differentiation into monocyte-like phenotype at concentrations much lower than 1 α ,25(OH)₂D₃. In contrast, we demonstrated that its epimer, (20*S*,23*R*)-epoxymethano-1 α ,25-dihydroxyvitamin D₃ (AMCR277B, **2b**, C23*R*, Figure 1), behaves like 1 α ,25(OH)₂D₃. Previous studies have shown that incorporation of an α -methyl at C2 of the A-ring of 1 α ,25(OH)₂D₃ does not result in superagonistic activity, although this compound (**3**, Figure 1) exhibits a 3-fold higher affinity to the VDR LBD relative to the natural ligand.^{20,21} In order to improve the capacity of 1 α ,25(OH)₂D₃ agonists in stimulating transcription and cellular growth arrest, we undertook the structure-based design of novel 1 α ,25(OH)₂D₃

[†]The atomic coordinates and structure factors (PDB codes 3A3Z and 3A40) have been deposited in the Protein Data Bank, Research Collaboratory for Structural Bioinformatics, Rutgers University, New Brunswick, NJ (<http://www.rcsb.org/>).

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^aAbbreviations: NR, nuclear receptor; 1 α ,25(OH)₂D₃, 1 α ,25-dihydroxyvitamin D₃; LBD, ligand binding domain; VDR, vitamin D receptor; rmsd, root-mean-square deviation; (S)-p-tolBINAP, (S)-(-)-2,2'-bis(di-*p*-tolylphosphino)-1,1'-binaphthyl.

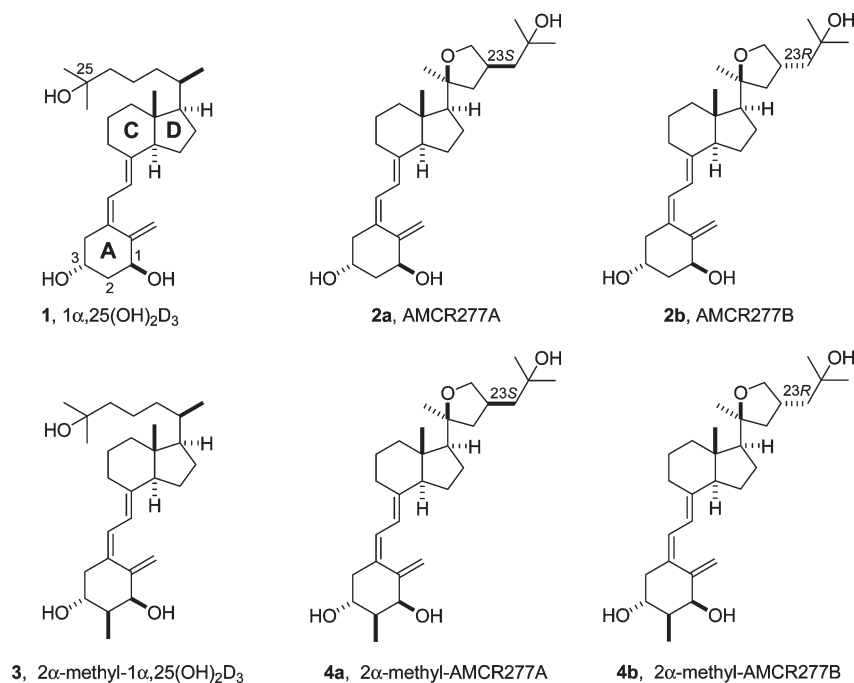
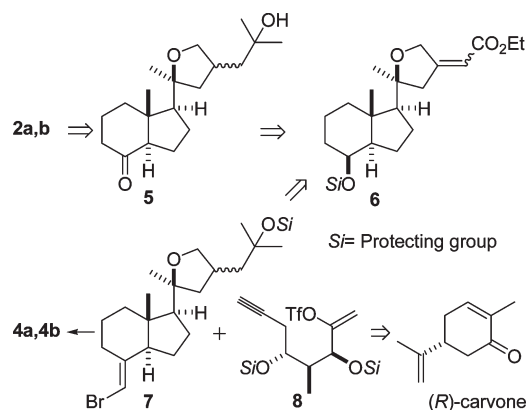


Figure 1. Chemical structures of the $1\alpha,25(\text{OH})_2\text{D}_3$ and the analogues **2a**, **2b**, **4a**, and **4b**.

Scheme 1. Synthesis of Vitamin D_3 Analogues **2a** and **2b** by the Wittig–Horner Approach



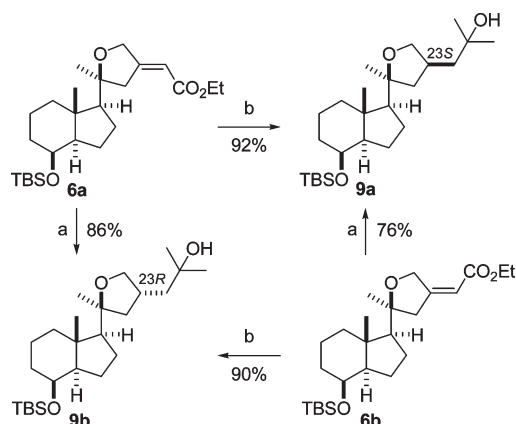
analogues, the (20*S*,23*S*)-epoxymethano- $1\alpha,25$ -dihydroxyvitamin D_3 (2 α -methyl-AMCR277A, **4a**, C23*S*, Figure 1) and its epimer, the (20*S*,23*R*)-epoxymethano- $1\alpha,25$ -dihydroxyvitamin D_3 (2 α -methyl-AMCR277B, **4b**, C23*R*, Figure 1). These new $1\alpha,25(\text{OH})_2\text{D}_3$ analogues were synthesized to increase their stability in the ligand binding pocket (LBP) to enhance $1\alpha,25(\text{OH})_2\text{D}_3$ superagonistic activity.

The present study aims at gaining insight into the structure–activity relationships of **4a,b**. To achieve this goal, we solved the crystal structures of the hVDR LBD bound to **4a** or to its epimer **4b** and highlight a sharp difference between the *in vitro* and the *in vivo* biological effects of the two $1\alpha,25(\text{OH})_2\text{D}_3$ analogues.

Chemistry

We have recently described the synthesis of active $1\alpha,25(\text{OH})_2\text{D}_3$ analogues **2a** and **2b** (Figure 1) by Lythgoe's Wittig–Horner approach using ketones **5** as the CD side chain building block. These ketones were prepared by catalytic hydrogenation of unsaturated esters **6** followed by the following sequence:

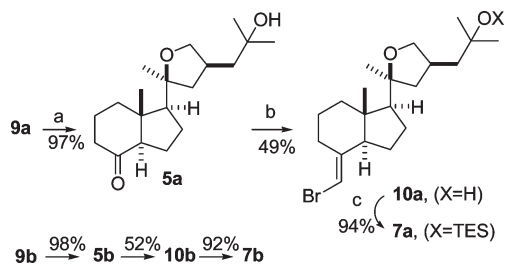
Scheme 2. Asymmetric Reduction of Unsaturated Esters **9a** and **9b**^a



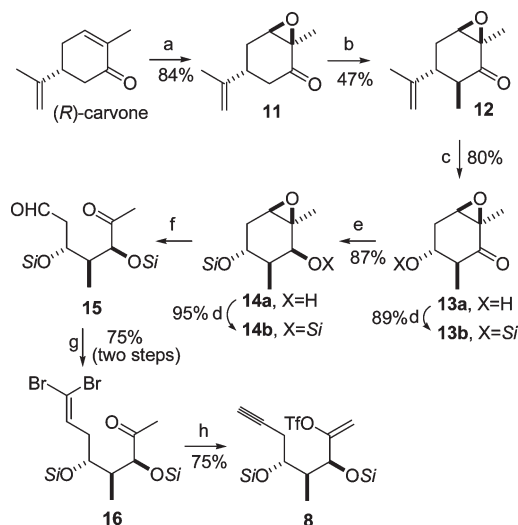
^a Reagents and conditions: (a) PMHS, CuCl, KO*t*-Bu, (*S*)-tolBINAP, *i*-PrOH, hexanes, -25°C , ultrasound, 1 h; then MeLi, THF, -78°C \rightarrow room temp; (b) same as in step a but with (*R*)-tolBINAP instead of (*S*)-tolBINAP.

methylation, separation of the resulting tertiary alcohols, desilylation, and oxidation (Scheme 1).¹⁹ We now report the synthesis of $1\alpha,25(\text{OH})_2\text{D}_3$ analogues **4a** and **4b** from (*R*)-carvone (A-ring precursor) and involving a stereoselective route to the CD-building block. The key features of the new synthetic strategy involve (1) palladium-catalyzed coupling between enoltriflate **8** and alkenylzinc intermediates derived from **7**, (2) stereoselective synthesis of bromides **7** from **6**, and (3) synthesis of enoltriflate **8** from (*R*)-carvone.

Synthesis of Alkenyl Bromides 7. Separation of a 1:1 mixture of (*E,Z*)-unsaturated esters **6**¹⁹ by HPLC provided pure (*E*)-ester (**6a**, less polar compound) and (*Z*)-ester (**6b**, more polar compound) (Scheme 2). The stereochemistry of both isomers was established by NOE experiments ($=\text{CHCO}_2\text{Et}$). Asymmetric conjugate reduction of unsaturated ester **6a** under Buchwald conditions²² employing catalytic

Scheme 3. Synthesis of the Upper Alkenyl Bromides **7a** and **7b**^a

^a Reagents and conditions: (a) TBAF, THF, reflux, 4 days; PDC, CH₂Cl₂, room temp, 12 h; (b) (Ph₃PCH₂Br)Br, NaHMDS, THF; (c) Et₃SiCl (TESCl), Im, DMAP, DMF, 0 °C → room temp, 12 h.

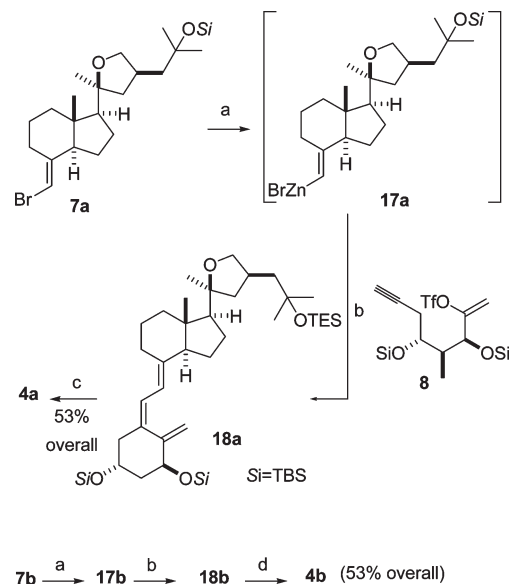
Scheme 4. Synthesis of the A-Ring Fragment^a

^a Reagents and conditions: (a) H₂O₂, LiOH·H₂O, MeOH, 0 °C, 30 min; (b) NaH, THF, room temp, 1 h; MeI, 0 °C, 5 h (47%); (c) O₃, MeOH-CH₂Cl₂, -78 °C; Ac₂O, Et₃N, DMAP, -35 °C → 78 °C, 2 h; NaOAc, MeOH, 37 °C, 12 h; (d) TBSCl (*t*-BuMe₂SiCl), imidazole, DMF, room temp, 12 h; (e) L-Selectride, THF, -78 °C, 30 min; (f) H₃IO₆, Et₂O, room temp, 12 h; (g) Ph₃P, Zn, CBr₄, CH₂Cl₂, 45 min; (h) LDA, THF, -78 °C, 1 h; *n*-BuLi, 15 min.; 5-Cl-py-2-NTf₂, -78 °C → room temp, 12 h.

(*R*)-*p*-tolBINAP/CuCl/KO*t*-Bu and poly(methylhydrosiloxane) (PMHS) as the reducing agent and *i*-PrOH as the proton source provided (C23*S*)-alcohol **9a**¹⁹ in 92% yield. The use of the enantiomeric ligand (*S*)-*p*-tolBINAP gave (C23*R*)-alcohol **9b**¹⁹ in 86% yield. This strategy allows the preparation of **9a** (precursor of superagonist **2a**) or **9b** as the only products from either unsaturated esters **6a** or **6b**. The above synthetic route allows the preparation of alcohol **9a** or **9b** in high yield from either ester **6a** or **6b**. The previous synthetic route¹⁹ provides a mixture of alcohols **9a** and **9b**.

Alcohols **9a** and **9b** were converted to the corresponding ketones **5a** and **5b** as previously described.¹⁹ Wittig olefination of **5a** and **5b** by the method of Trost²³ followed by silylation provided the required alkenyl bromides **7a** and **7b** in 46% and 48% yield, respectively (two steps) (Scheme 3).

Synthesis of A-Ring Fragment 8. The synthesis of enoltriflate **8**, precursor of the A-ring, started with (*R*)-carvone²⁴ (Scheme 4). Stereoselective epoxidation of (*R*)-carvone provided the known epoxide **11**^{24g,25} in 84% yield. Treatment of **11** with sodium hydride and alkylation with methyl iodide gave the monoalkylated product **12** in 47% yield, together

Scheme 5. Formation of the Triene System by Pd(0)-Catalyzed Tandem Cyclization Negishi Coupling^a

^a Reagents and conditions: (a) *t*-BuLi, THF, -78 °C, 1 h; ZnBr₂, THF, -78 °C → 0 °C, 1 h; (b) **8**, (Ph₃P)₄Pd, Et₃N, THF, 12 h; (c) *n*-Bu₄NF, THF, room temp, 24 h; (d) HF·Py, CH₂Cl₂, CH₃CN, Et₃N, room temp, 2 h.

with the corresponding dialkylated compound (29%). The stereochemistry of the methylation product **12** can be explained by protonation of the corresponding enolate. Degradation of the isopropenyl side chain^{25b} of **12** by ozonolysis and subsequent reaction of the resulting intermediates with acetic anhydride followed by treatment with sodium acetate provided the desired alcohol **13a** in 80% yield. Silylation of **13a** with *tert*-butyldimethylsilyl chloride followed by reduction of the resulting silylated compound **13b** with lithium selectride followed by protection afforded the disilylated epoxide **14b** (77% yield for the three steps). The stereochemical outcome of the reduction product **14a** can be explained by L-Selectride attack to ketone **13b** from the less hindered face. The structure of compound **14a** was established by NOE experiments (see Supporting Information). Periodic acid-oxidative cleavage of **14b** generated the aldehyde **15** which was transformed into the vinylic dibromide **16** by olefination with Ph₃P=CBr₂ (75% yield, 2 steps). Conversion of **16** to the desired enoltriflate **8** was carried out in 75% yield using conditions previously reported for the corresponding unalkylated compound:²⁴ⁱ base treatment (LDA and *n*-BuLi) to form the triple bond and the enolate and trapping of the latter with *N*-(5-chloro-2-pyridyl)bis(trifluoromethanesulfonamide).

Synthesis of the Target 1α,25(OH)₂D₃ Analogues 4a and 4b. For the construction of the vitamin D triene system, we chose the convergent Pd(0)-catalyzed tandem cyclization Negishi coupling approach recently developed in our laboratories.^{24k} The alkenylzinc intermediate **17a** (CD side chain building block, Scheme 5) was prepared by metalation of alkenyl bromide **7a** with *t*-BuLi followed by transmetalation with ZnBr₂. Addition of the enoltriflate **8**, triethylamine, and catalytic tetrakis(triphenylphosphine)palladium(0) gave, after desilylation, the desired analogue **4a** in 55% yield. The other analogue **4b** was prepared in a similar way (53% yield) from alkenyl bromide **7b**.

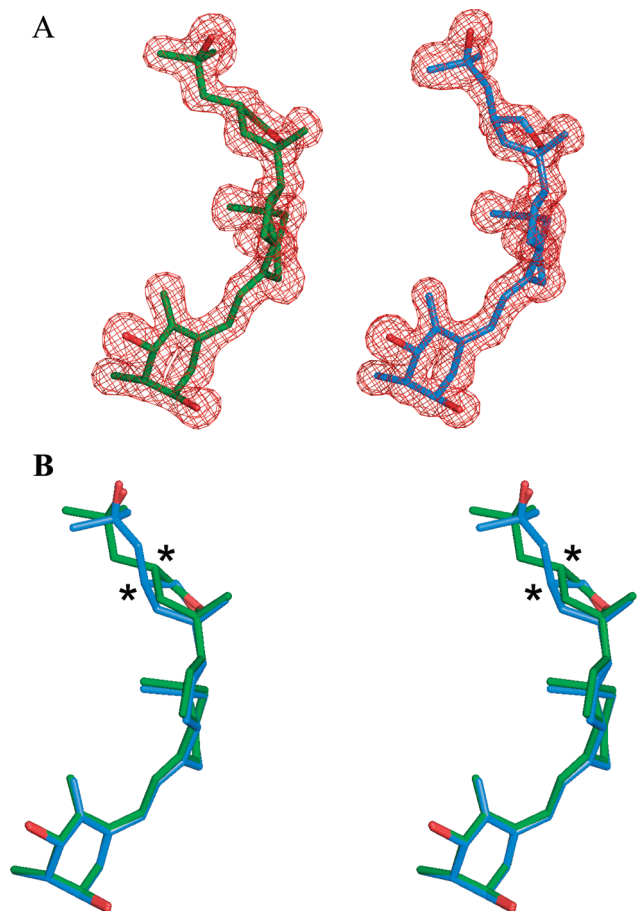


Figure 2. Conformation of the bound ligand. (A) The **4a** (left) and **4b** (right) are shown in the $F_o - F_c$ electron density omit map contoured at 3σ . The ligands are shown in stick representation with carbon atoms in green for **4a** and blue for **4b**, and oxygen atoms are in red. (B) A stereoview of the ligand conformations of **4a** (green) and **4b** (blue) in the VDR ligand binding pocket after superimposition showing the different side chain conformation of the two epimers. The chiral atoms at C23 are marked by asterisks.

Results

Overall Structures of the hVDR LBD Bound to **4a** or **4b**.

The hVDR LBD mutant lacking 50 residues in the loop connecting helices H2 and H3 was used for the X-ray analyses of the hVDR LBD in complex with **4a** or **4b**. The same mutant was used to solve the structure of the hVDR LBD bound to $1\alpha,25(\text{OH})_2\text{D}_3$ and to several $1\alpha,25(\text{OH})_2\text{D}_3$ analogues, since the biological properties of this mutant such as ligand binding and transactivation in distinct cell lines are the same as those of the wild type.^{6,26–28} Isomorphous crystals were obtained in similar conditions, and the crystal structures of hVDR LBD bound to **4a** or **4b** were determined at a resolution of 1.7 and 1.45 Å, respectively (Supporting Information Table 1).

Complexes of hVDR LBD with **4a** or **4b** adopt the canonical conformation of all previously reported agonist-bound nuclear receptor LBDs with 12 or 13 α -helices organized in a three-layered sandwich. In all the structures of hVDR LBD bound to agonist ligands, a unique conformation of the complex is observed. The position and conformation of the activation helix H12 are strictly maintained. The ligand adopts the same orientation in the pocket. An adaptation of their conformation is observed to maintain the

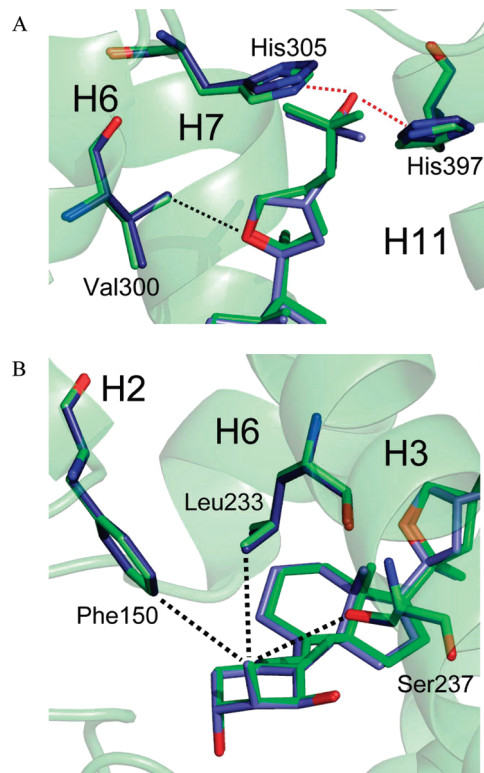


Figure 3. Detailed structural representation of both analogues bound to hVDR LBD around the aliphatic chain (A) and the 2α -methyl group (B) showing the characteristic residues involved in interactions. **4a** and **4b** are shown in green and blue, respectively. Oxygen and nitrogen atoms in VDR/**4a** and VDR/**4b** structures are shown in red and blue, respectively. Hydrogen and van der Waals bonds are shown in red and black dotted lines, respectively. Secondary structure of VDR is shown in cartoon. H2, H3, H6, H7, and H11 indicate helices.

hydrogen bonds forming the anchoring points. Compared to the structure of hVDR LBD in complex $1\alpha,25(\text{OH})_2\text{D}_3$, the atomic coordinates of all C α atoms of hVDR bound to **4a** and to **4b** show an rmsd of 0.18 and 0.15 Å, respectively. The ligand is buried in the predominantly hydrophobic pocket. The volumes of the ligands are 418, 416, and 396 Å³ for **4a**, **4b**, and $1\alpha,25(\text{OH})_2\text{D}_3$, respectively. The volume of the ligand binding cavity is 685, 682, and 673 Å³ and the ligands occupy 61%, 61%, and 59% of the pocket for **4a**, **4b**, and $1\alpha,25(\text{OH})_2\text{D}_3$, respectively.

Ligand–Protein Interactions. The A, seco-B, C, and D rings present conformations that are similar to those observed in the presence of the natural ligand (Figure 2). The distances between the 1-OH and the 25-OH groups are 12.9, 13.1, and 12.8 Å for hVDR LBD bound to **4a**, **4b**, and $1\alpha,25(\text{OH})_2\text{D}_3$, respectively. The hydroxyl groups make the same hydrogen bonds as hVDR LBD bound to $1\alpha,25(\text{OH})_2\text{D}_3$ complex, 1-OH with Ser237 and Arg274, 3-OH with Tyr143 and Ser278, and 25-OH with His305 and His397.

A comparison of the aliphatic chain of **4a** to those of the previous nonmethylated **2a** and **2b**¹⁹ reveals that the specific van der Waals interactions of the side chain with His-305, His-397, Val-418 and of the O21 atom of the oxolane ring with Val300 are conserved (Figure 3A). Similar to **2b**, the additional van der Waals contact of O21 with Val300 is weaker in the VDR–**4b** complex. This interaction with Val300 has been observed in some hVDR LBD crystal

structures with superagonist ligands $1\alpha,25$ -dihydroxy-20-*epi*-22-*oxa*-24a,26a,27a-trihomovitamin D_3 (KH1060) or 19-nor-14-*epi*-23-yne-1,25-(OH) $_2D_3$ (TX522).^{28,29} The stereoisomer **4b** presents an energetically unfavorable oxolane ring and adopts a different side chain conformation because of the inverse configuration at C23 which affects the positions of C25, C26, C27, and 25-OH and destabilizes the activation helix-12 (Val418) compared to **4a**.

The introduction of a methyl group at the 2α position does not modify the A-ring chair conformation of each ligand. The 2α -methyl group fills a small cavity of the pocket and makes additional van der Waals interactions with Phe150, Leu233, and Ser237 (Figure 3B), contacts also observed in the hVDR LBD complexed to 2α -methyl- $1\alpha,25$ (OH) $_2D_3$.²⁰ The previously reported crystal structures of hVDR LBD in complex with $1\alpha,25$ (OH) $_2D_3$ and several synthetic ligands revealed the presence of tightly bound water molecules forming a channel near the C2 position of the ligand, which may play important roles in protein stability.²⁰ This water channel is also conserved in the two present **4a** and **4b** complexes, and the *B*-factors of the three waters (7.0, 6.4, and 3.3 Å² for the hVDR LBD/**4a** complex and 12.4, 11.9, and 10.3 Å² for the **4b** complex, respectively) are significantly lower than the average value of all water molecules (28.1 and

32.3 Å² for the hVDR LBD/**4a** complex and **4b** complex, respectively).

Binding Affinity of the SRC-1 Peptide to hVDR LBD Bound to **2a or **4a** Is Similar.** To check the ligand effect on the recruitment of coactivator peptide to VDR, binding of fluorescently labeled SRC-1 peptide to hVDR LBD bound to $1\alpha,25$ (OH) $_2D_3$, **2a**, **2b**, **4a**, and **4b** was monitored by fluorescence anisotropy. In its apo form, hVDR LBD does not bind the SRC-1 coactivator peptide. The magnitude of the SRC-1 binding when complexed to **2a** or **4a** was similar to that obtained with the natural ligand. In contrast, the **4b** ligand induces a 3-fold weaker binding between hVDR LBD and the SRC-1, whereas its parental analogue **2b** lowers it by 1.7-fold (Table 1). Therefore, stereochemistry of the C23 is a significant parameter for the recruitment of the SRC-1 peptide.

4a Is a More Potent $1\alpha,25$ (OH) $_2D_3$ Superagonist Than **2a.** Using gene reporter assays, we recently reported that **2a** shows superagonistic biological activities because it stimulates transcription 12-fold more than the natural ligand at 0.1 nM.¹⁹ Methylation at the C2 α position of the $1\alpha,25$ (OH) $_2D_3$ A ring induces a 3-fold increase of the binding affinity of hVDR LBD. Hence, we postulate that C2 α methylation of **2a** may further increase its superagonist character.

Here, we demonstrate that at 0.1 nM, the transcription activity upon **4a** stimulation is 13- and 3.8-fold higher than those induced by the natural ligand and **2a**, respectively. Furthermore, at 5 nM **4a** achieves optimal VDR-directed transcription, whereas **4b** is 4 times less potent under similar stimulatory conditions in transactivation assays and behaves like the natural ligand (Figure 4). The dose-dependent comparison between **2a** and its new to 2α -methyl derivative reveals that at 0.1, 1, and 5 nM, the transcription activities induced by **2a** are only 26%, 42%, and 73% of that obtained

Table 1. Dissociation Constants of the SRC-1 Peptide to hVDR LBD in Response to $1\alpha,25$ (OH) $_2D_3$ Analogues

analogues	relative dissociation constants
no ligand	no binding
$1\alpha,25$ (OH) $_2D_3$	1
2a	0.9 ± 0.2
2b	1.7 ± 0.2
4a	1.3 ± 0.2
4b	3.2 ± 0.2

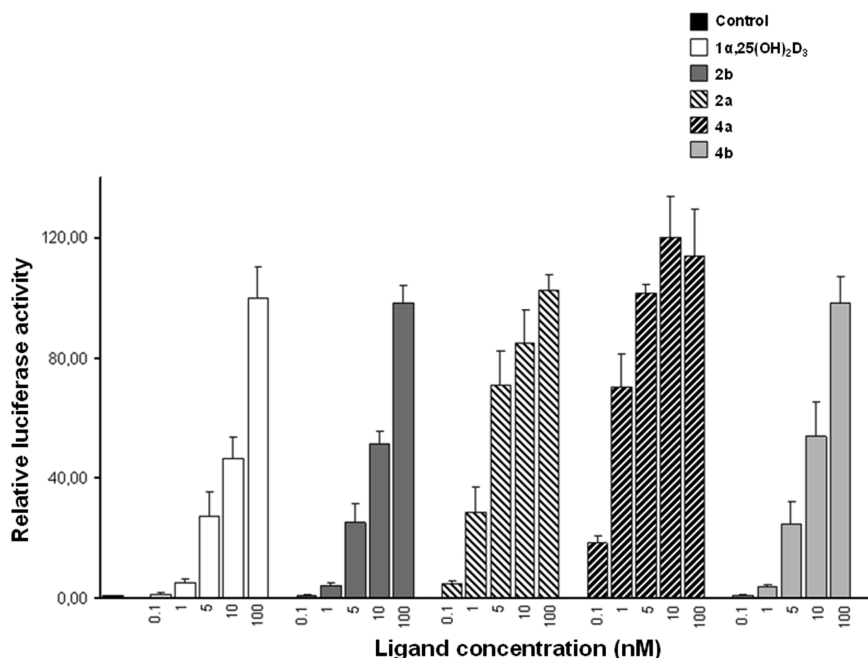
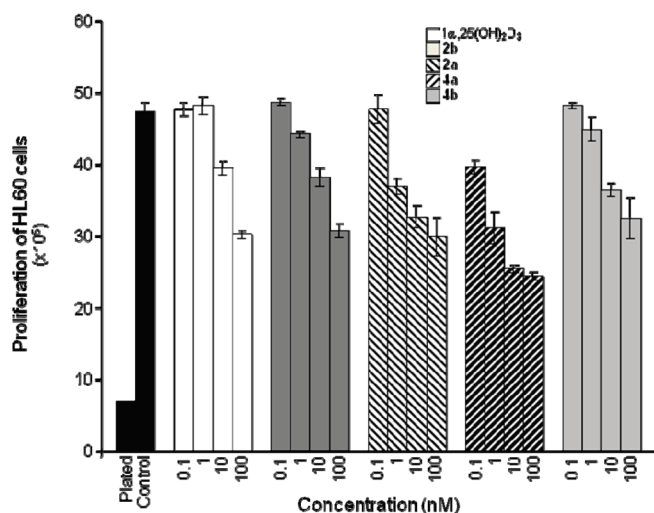


Figure 4. **4a** acts as an $1\alpha,25$ (OH) $_2D_3$ superagonist in vitro. Transient transfections and luciferase reporter gene assays using MCF-7 human breast cancer cells were performed as reported in Experimental Procedures. Relative luciferase activities were calculated by dividing the luciferase activity by the respective β -galactosidase activity to correct for differences in transfection efficiency. For every triplicate the mean and the standard deviation were calculated. A two-tailed, paired Student's *t* test was performed when appropriate, and *p* values were calculated with reference to stimulation of hVDR with the solvent.

A



B

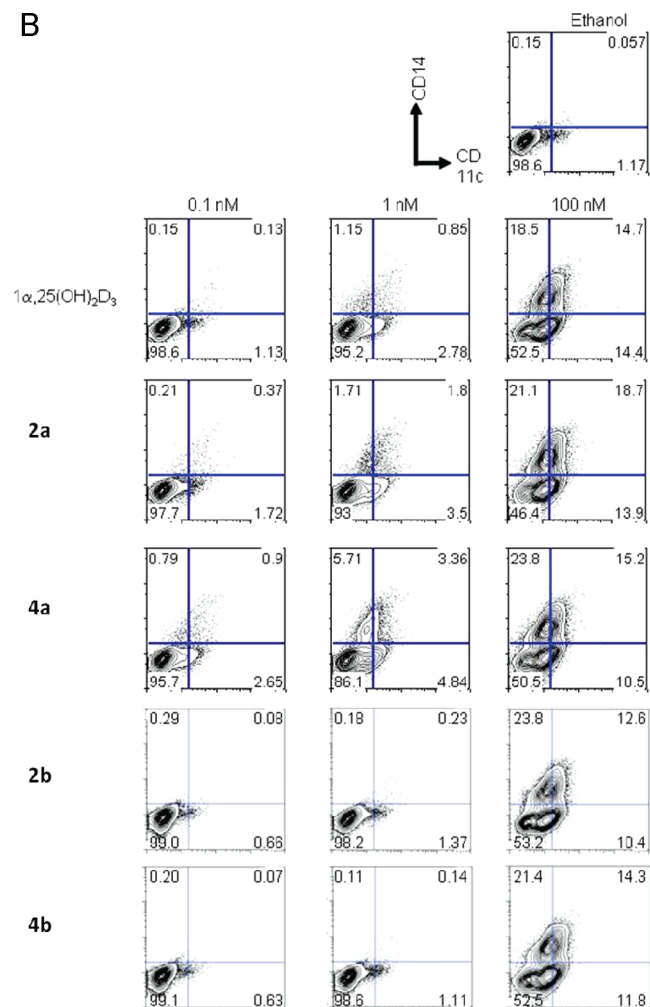


Figure 5. **4a** displays higher HL60 cell antiproliferation and prodifferentiating potency relative to its parental analogue **2a**. (A) **4a** mediated HL60 cell growth arrest is achieved for a dose of 0.1 nM. HL60 cells were incubated for 96 h with various concentrations of $1\alpha,25(\text{OH})_2\text{D}_3$, **2a**, **2b**, **4a**, or **4b** versus control (ethanol 0.7%) and are counted. Data are presented as the mean \pm SEM. (B) **4a** mediated HL60 differentiation into monocyte-like cells is achieved for a dose of 0.1 nM. Cells were labeled with PE-labeled antihuman CD11c and FITC-labeled antihuman CD14, and HL60 cell differentiation was estimated by the double-positive CD11c/CD14 subpopulation. Data are representative of three distinct experiments.

in the presence of **4a**, respectively. In summary, **4a** is a more potent superagonist compared to **2a** while its epimer **4b** is as active as the natural ligand (**1**).

Only 4a at 0.1 nM Induces HL60 Cell Proliferation Arrest with Concomitant Differentiation. Our previous data showed that the superagonistic property of **2a** is associated with a higher ability, compared to the natural ligand, to inhibit the proliferation of HL60 promyelocytic cell and to trigger their subsequent differentiation into monocyte-like phenotype as evidenced by the up-regulation of cell surface markers CD11c and CD14.¹⁹ We further examined the effects of **4a** and **4b** in directing HL60 cell fate. Here, we found that the dose-dependent profile of **4b** in mediating HL60 cell arrest and differentiation is similar to that induced by **2b**. In sharp contrast, a dose of only 0.1 nM of **4a** is able to significantly reduce HL60 cell proliferation, whereas no effect at this concentration was observed for the other $1\alpha,25(\text{OH})_2\text{D}_3$ analogues, including its parental analogue **2a**, compared to control incubations (Figure 5A). A similar proliferation pattern was detected when HL60 cells were incubated with 1 nM $1\alpha,25(\text{OH})_2\text{D}_3$ agonists. Consistent with cell proliferation data, a 10-fold gain in the induction of HL60 cells

differentiation was obtained with 0.1 nM **4a**, as a new double positive CD11c/CD14 subpopulation was detected compared to the other $1\alpha,25(\text{OH})_2\text{D}_3$ analogues (Figure 5B). Moreover, single positive CD11c and CD14 cell populations were more abundant relative to **2a** and the natural ligand (at least 1.5- and 3-fold increase, respectively). A similar profile was observed with 10 nM $1\alpha,25(\text{OH})_2\text{D}_3$ analogues treatment, and a lower or equal dose of 2 α -methyl- $1\alpha,25(\text{OH})_2\text{D}_3$ achieves neither antiproliferation nor differentiation of HL60 cells (data not shown). Therefore, C2 α methylation of **2a** increases its ability to inhibit HL60 cell proliferation with concomitant differentiation into monocyte-like phenotype.

At low dose, 4a is more calcemic than $1\alpha,25(\text{OH})_2\text{D}_3$. Whereas 4b Displayed in Vivo Low Calcemic Action Compared to the Natural Ligand. Using C57BL/6J mice, we previously showed that **2a** exhibits higher calcemic action than $1\alpha,25(\text{OH})_2\text{D}_3$, whereas **2b** has an in vivo low calcemic effect than the natural ligand.¹⁹ We monitored the calcium serum level of male mice that were subjected to 0.1 $\mu\text{g}/\text{kg}$ intraperitoneal injections (ip) of $1\alpha,25(\text{OH})_2\text{D}_3$ analogues versus untreated (control) ones. Mice survive during the

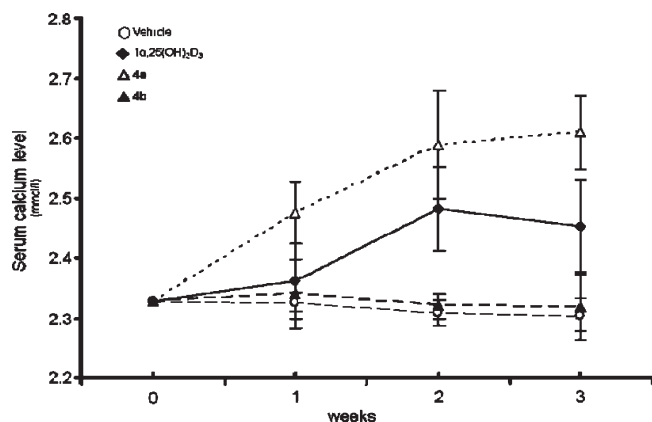


Figure 6. **4a** is more calcemic than **2a** or $1\alpha,25(\text{OH})_2\text{D}_3$, whereas **4b** is less calcemic. Mice (6 weeks old; 6–7 mice/group) were injected intraperitoneally with $1\alpha,25(\text{OH})_2\text{D}_3$, **4a**, **4b**, or sesame oil (vehicle) for 3 weeks. A single dose of $0.1 \mu\text{g}/\text{kg}$ was administered every second day, and calcium serum level was weekly monitored: (open diamond) control, (black diamond) $1\alpha,25(\text{OH})_2\text{D}_3$, (open triangle) **4a**, and (black triangle) **4b** incubations. Data are presented as the mean \pm SEM.

3 weeks of the experiments, and their physiological state was satisfactory in response to the different analogues (data not shown). We observed that mice treated with **4b** leads to calcemic values that are similar to those induced by sesame oil as vehicle incubations. In contrast, ip injections of **4a** promote a robust calcium increase in the serum of mice, which is even higher than that detected with the natural ligand and similar to **2a**¹⁹ (Figure 6). Thus, C2 α methylation of **2a** improves the superagonist character of its parental analogue but not its calcemic property.

Discussion

The present study reports the crystal structures of the human VDR LBD bound to novel $1\alpha,25(\text{OH})_2\text{D}_3$ analogues, the **4a**, and its epimer **4b** and describes their in vitro and in vivo biological effects. The rationale to undertake this study was based on three distinct observations: (i) compared to the crystal structure of hVDR LBD in complex with its natural ligand, those with 2α -methyl- $1\alpha,25(\text{OH})_2\text{D}_3$ and **2a** show different van der Waals interactions that may explain in part their biological action; (ii) 2α -methyl- $1\alpha,25(\text{OH})_2\text{D}_3$ displays higher binding affinity for hVDR LBD compared to $1\alpha,25(\text{OH})_2\text{D}_3$; (iii) **2a** is a $1\alpha,25(\text{OH})_2\text{D}_3$ superagonist or has the capacity to better induce the differentiation of HL60 cells into monocyte-like phenotype than the natural ligand, while its epimer behaves like the natural ligand.^{19,20} Thus, a methylation at the C2 α of the A ring combined with an incorporation of an oxolane ring in the side chain leads to a new $1\alpha,25(\text{OH})_2\text{D}_3$ analogue that may behave as a more potent superagonist associated with stronger antitumoral properties in vitro. Therefore, we synthesized the **4a** and its epimer **4b** (Figure 1) and investigated their structure–function relationships.

Atomic resolution of the crystal structure of the hVDR LBD bound to **4a** reveals that this complex inherits structural features of both 2α -methyl- $1\alpha,25(\text{OH})_2\text{D}_3$ and **2a**, as evidenced by the specific van der Waals contacts of the 2α -methyl group with Phe150, Leu233, Ser237 and of the O21 atom of the oxolane ring with Val300 (Figure 3). As previously reported for hVDR LBD bound to agonist ligands, helix 12

adopts the canonical orientation, sealing the ligand binding cavity.^{22,30,31} Comparison of complexes formed by hVDR LBD with **2a**, $1\alpha,25(\text{OH})_2\text{D}_3$ or 2α -methyl- $1\alpha,25(\text{OH})_2\text{D}_3$, indicates that the new ligand **4a** displays an increased number of van der Waals contacts with the LBP than the other ones. These additional contacts may explain in part the higher potency of **4a** in mediating VDR-dependent transcription. Although 2α -methyl- $1\alpha,25(\text{OH})_2\text{D}_3$ displays a higher binding affinity for hVDR LBD relative to its natural ligand, it does not exhibit superagonist properties. On the other hand, we have shown that **2a** displays superagonistic biological activities.¹⁹ Therefore, methylation at the C2 α of the **2a** analogue synergistically increases its superagonistic character (Figure 4). In contrast, C2 α methylation of **2b** does not improve its ability to further induce VDR-directed transcription compared to its parental **2b** analogue or the natural ligand (Figures 2 and 3). Because of the inverse configuration at C23, the stereoisomers **4b** and **2a** adopt a different side chain conformation compared to **4a** and **2a** affecting their respective contacts with activation helix 12 and resulting in a weaker activation. This property is conserved in the **4b** relative to **4a**, demonstrating that the active form of the bound ligand favored by the ring pucker is an essential parameter that directs the magnitude of VDR-induced transcription and HL60 cell anti-proliferation/prodifferentiation (Figures 4 and 5).

Previous studies have shown that stronger VDR–coactivator interactions underlie the superagonistic activity of $1\alpha,25(\text{OH})_2\text{D}_3$ analogues, such as 14-epi analogues.²⁸ To gain insights into the potential molecular mechanisms that may explain the superagonistic strength of **4a** relative to **2a**, we determined the binding affinity of the coactivator SRC-1 peptide to hVDR LBD incubated with each analogue. A higher recruitment of SRC-1 to hVDR LBD upon binding to **4a** does not account for its increased superagonist character, as **4a** induces weaker binding between hVDR LBD and the SRC-1 peptide compared to either $1\alpha,25(\text{OH})_2\text{D}_3$ or **2a** which induces a binding efficiency fully comparable to that obtained with the natural ligand (Table 1). Therefore, larger domains of SRC-1 or other coactivators may play a key role in VDR-dependent transcription in response to **4a**.

Since the **2a** superagonist character is associated with a more potent ability to inhibit HL60 cell proliferation with concomitant differentiation into monocyte-like cells, we postulate that increased stabilization of the complex hVDR LBD bound to **4a** would result in a higher capacity to reduce HL60 cell growth arrest at low concentrations. We recapitulated the profile of each previously described $1\alpha,25(\text{OH})_2\text{D}_3$ analogue in the modulating HL60 cell fate, namely, **2a** at 1 nM induces HL60 cell antiproliferation, whereas **2b** behaves like the natural ligand.¹⁹ We now show that **4a** induces a 10-fold gain in lowering HL60 cell proliferation, as 0.1 nM **4a** leads to a significant decrease of HL60 cells number, whereas no effect was observed at this concentration with the other $1\alpha,25(\text{OH})_2\text{D}_3$ analogues, including **4b** (Figure 5A). A similar profile is observed at a dose of 1 nM, **4a** being more efficient compared to its parent analogue or 2α -methyl- $1\alpha,25(\text{OH})_2\text{D}_3$ (Figure 5A and not shown). Consistent with our previous findings, the antiproliferative effect of **4a** correlates with its capacity to better differentiate HL60 cells into monocyte-like cells compared to **2a** (Figure 5B). It should be emphasized that the 2α -methyl- $1\alpha,25(\text{OH})_2\text{D}_3$ has been reported to be between 2- and 4-fold more potent relative to $1\alpha,25(\text{OH})_2\text{D}_3$ in inducing differentiation of HL60 cells, depending on the quantified parameter (nitroblue tetrazolium reduction assay

and CD11b surface cell marker expression).^{21,32,33} However, no information of the concentration of the analogues used in these studies was provided. Our present data indicate that 1 nM 2 α -methyl-1 α ,25(OH)₂D₃ was not sufficient to induce HL60 cells antiproliferation or differentiation (not shown). However, when used at the same low concentration, the C2 α -methylated derivative of **2a** does achieve HL60 cell differentiation. This property is associated with a better binding affinity of **4a** for VDR compared to **2a** and 1 α ,25(OH)₂D₃, as evidenced by electrospray ionization mass spectrometry experiments (data not shown).

The in vitro biological data were not accompanied with a better capacity to reduce the serum calcium level in **4a**-treated mice. Nevertheless, we report that ip administration for 3 weeks of **4a** or **4b** every 2 days results in an optimal survival of mice, as no mice death was observed. Further dissociation between antiproliferative and calcemic effects has to be improved.

In summary, we have synthesized a novel 1 α ,25(OH)₂D₃ analogue, the 2 α -methyl-20*S*,23*S*-epoxymethano-1 α ,25-dihydroxyvitamin D₃ (**4a**), which exhibits superagonistic and cellular antiproliferation/prodifferentiation properties while showing similar calcemic effects. In addition, its epimer **4b** (23*R*), which behaves like the natural ligand in vitro, is a noncalcemic vitamin D in vivo. To validate the potential application of these two analogues, further in vitro studies in distinct cell types are required to determine the specific gene activity and metabolic degradation.

Experimental Procedures

Chemistry. General. For general experimental procedures, see ref 19. CD-ring intermediates were named following the steroidal nomenclature.³⁴ Vitamin D analogues are named as vitamin D derivatives. IUPAC rules were used for the other compounds. In addition to NMR, HPLC analysis was used to determine the purity (>95%) of the vitamin D analogues.

Ethyl 8 β -[(*tert*-Butyldimethylsilyloxy)-(20*R*,23)-epoxymethano-*des-A,B*-cholest-23*E*-en-25-oate (6a**) and Ethyl 8 β -[(*tert*-Butyldimethylsilyloxy)-(20*R*,23)-epoxymethano-*des-A,B*-cholest-23*Z*-en-25-oate (**6b**).** A mixture of **6a** and **6b** (1.01 g), prepared by phosphonate chemistry,¹⁹ was separated by HPLC (Phenomenex Silica(2), 250 mm \times 21.2 mm, isocratic 3% EtOAc/hexanes) to give **6a** (*E*, 0.52 g, 46%, colorless oil) and **6b** (*Z*, 0.48 g, 42%, white solid, mp = 101 °C). Spectral data for the compounds **6a** and **6b** were identical to those reported.¹⁹

8 β -[(*tert*-Butyldimethylsilyloxy)-(20*S*,23*S*)-epoxymethano-25-hydroxy-*des-A,B*-cholestane (9a**) from **6a** and **6b**.** PMHS [(poly(methylhydrosiloxane))] (0.11 mL, 1.82 mmol, 16 equiv) was added to a suspension of CuCl (8 mg, 0.08 mmol, 0.7 equiv), KO*t*-Bu (10 mg, 0.086 mmol, 0.75 equiv), and (*R*)-*p*-tolBINAP (9.3 mg, 0.014 mmol, 0.12 equiv) in hexanes (2 mL). The mixture was sonicated for 2 h. The color changed from yellow to deep-red. After the mixture was cooled to -25 °C, a solution of **6a** (50 mg, 0.114 mmol, 1 equiv) in hexanes (2 mL) and *i*-PrOH (0.035 mL, 0.456 mmol, 4 equiv) was added. After 1 h, the reaction was quenched by the addition of saturated NH₄Cl (5 mL) and EtOAc (5 mL). The mixture was extracted with EtOAc (2 \times 10 mL), and the combined organic phase was dried, filtered, and concentrated in vacuo. The residue was dissolved in dry THF and cooled to -78 °C. After 15 min, a solution of MeLi in Et₂O (0.18 mL, 0.285 mmol, 1.6 M, 2.5 equiv) was added via syringe. The resulting solution was stirred for 5 min at -78 °C and then at room temp for 1 h. After the mixture was cooled to -78 °C, the reaction was quenched by the slow addition of saturated NH₄Cl (3 mL). The mixture was extracted with Et₂O (3 \times 5 mL). The combined organic phase was dried, filtered, and

concentrated in vacuo. The residue was purified by flash chromatography (SiO₂, 1.5 cm \times 9 cm, 10% EtOAc/hexanes) to give **9a**¹⁹ [45 mg, 92% (two steps), *R*_f = 0.2 (20% EtOAc/hexanes)]. Following a similar procedure, **9a** was also prepared from **6b** in 76% yield. Spectral data for the compound **9a** were identical to those reported.¹⁹

8*E*-Bromomethylene-(20*S*,23*S*)-epoxymethano-25-hydroxy-*des-A,B*-cholesta-8-one (10a**).** Sodium hexametildisilazide (0.75 mL, 1.5 mmol, 2 M, 3 equiv) was added to (bromomethylene)-triphenylphosphonium bromide (0.680 g, 1.56 mmol, 3.1 equiv) in 10 mL of THF at -60 °C. After 1 h, a solution of ketone **5a** (0.155 g, 0.502 mmol, 1 equiv) in THF (5 mL) was added via cannula. The solution was stirred at -60 °C for 5 min and then at room temperature for 3 h. Hexanes were added, and the suspension was filtered over a pad of silica gel (elution with 20% Et₂O/hexanes). After concentration in vacuo, the residue was purified by flash chromatography (SiO₂, 2 cm \times 12 cm, 10% EtOAc/hexanes) to give the bromide **10a** [94 mg, 49%, colorless oil, *R*_f = 0.6 (50% EtOAc/hexanes)]. ¹H NMR (250 MHz, CDCl₃): δ 5.61 (broad s, 1H, H-7), 4.09 (t, 1H, *J* = 7.9 Hz, H-28), 3.32 (t, 1H, *J* = 9.2 Hz, H-28), 2.83 (m, 1H, H-9), 2.45 (m, 1H, H-23), 1.22 (s, 3H, CH₃-21), 1.20 (s, 6H, CH₃-26 and CH₃-27), 0.66 (s, 3H, CH₃-18). ¹³C NMR (62.9 MHz, CDCl₃): δ 144.8 (=C, C-8), 97.5 (=CH, C-7), 84.8 (C, C-20), 75.0 (CH₂, C-28), 70.7 (C, C-25), 59.0 (CH, C-17), 56.0 (CH, C-14), 46.3 (CH₂), 46.0 (CH₂), 45.5 (C, C-13), 40.0 (CH₂), 34.1 (CH, C-23), 30.9 (CH₂), 30.0 (CH₃, C-27), 29.7 (CH₃, C-26), 27.2 (CH₃, C-21), 22.5 (CH₂), 22.3 (CH₂), 21.7 (CH₂), 13.3 (CH₃, C-18).

8*E*-Bromomethylene-(20*S*,23*S*)-epoxymethano-25-[triethylsilyloxy]-*des-A,B*-cholestan-8-one (7a**).** Triethylsilyl chloride (0.140 mL, 0.84 mmol, 3 equiv) was added dropwise to a solution of alcohol **10a** (0.108, 0.280 mmol, 1 equiv), imidazol (0.06 g, 0.84 mmol, 3 equiv), and DMAP (0.007 g, 0.06 mmol, 0.2 equiv) in dry DMF (2 mL) at 0 °C. The cooling bath was removed, and the mixture was stirred for 12 h and poured onto a mixture of ice and saturated NH₄Cl (10 mL). The aqueous phase was extracted with hexanes (3 \times 10 mL). The combined organic phase was dried, filtered, and concentrated in vacuo. The residue was purified by flash chromatography (SiO₂, 2 cm \times 10 cm, 5% EtOAc/hexanes) to give **7a** [0.132 g, 94%, solid, *R*_f = 0.66 (10% EtOAc/hexanes)]. IR (dissolved in CHCl₃, cm⁻¹): 1630 ($\nu_{C=C}$). ¹H NMR (250 MHz, CDCl₃): δ 5.63 (broad s, 1H, H-7), 4.09 (t, 1H, *J* = 7.9 Hz, H-28), 3.32 (dd, 1H, *J*₁ = 9.7 Hz, *J*₂ = 8.6 Hz, H-28), 2.84 (m, 1H, H-9), 2.47 (m, 1H, H-23), 2.05 (d, 1H, *J* = 12.5 Hz, H-12), 1.97 (dd, 1H, *J*₁ = 11.9 Hz, *J*₂ = 7.3 Hz, H-14), 1.22 (s, 3H, CH₃-21), 1.20 (s, 6H, CH₃-26 and CH₃-27), 0.93 (t, 9H, *J* = 7.8 Hz, 3 \times CH₃CH₂-Si), 0.68 (s, 3H, CH₃-18), 0.55 (c, 6H, *J* = 7.8 Hz, 3 \times CH₃CH₂-Si). ¹³C NMR (62.9 MHz, CDCl₃): δ 144.9 (=C, C-8), 95.6 (=CH, C-7), 84.4 (C, C-20), 75.2 (CH₂, C-28), 73.1 (C, C-25), 59.1 (CH, C-17), 56.0 (CH, C-14), 47.9 (CH₂), 46.2 (CH₂), 45.5 (C, C-13), 40.0 (CH₂), 34.2 (CH, C-23), 30.9 (CH₂), 30.7 (CH₃, C-27), 30.0 (CH₃, C-26), 27.2 (CH₃, C-21), 22.6 (CH₂), 22.4 (CH₂), 21.7 (CH₂), 13.4 (CH₃, C-18), 7.1 (3 \times CH₃, 3 \times CH₃CH₂-Si), 6.7 (3 \times CH₂, 3 \times CH₃CH₂-Si). HMRS ([CI]⁺): calculated for [C₂₆H₄₈O₂Si⁷⁹Br]⁺ [(M + H)⁺], 499.2607; found, 499.2609.

(1*R*,3*S*,4*R*,6*R*)-1,3-Dimethyl-4-(prop-1-en-2-yl)-7-oxa-bicyclo-[4.1.0]heptan-2-one (12**).** A solution of ketone **11** (10 g, 59.4 mmol, 1 equiv) in dry THF (50 mL) was added via syringe to a suspension of NaH (2.16 g, 89.1 mmol, 1.5 equiv) in THF (65 mL). After being stirred for 1 h, the mixture was cooled to 0 °C and then stirred for 10 min. MeI (4.5 mL, 72.3 mmol, 1.2 equiv) was added dropwise via syringe. The reaction mixture was stirred for 4 h. The reaction was quenched with saturated NH₄Cl (100 mL). The mixture was extracted with Et₂O (3 \times 50 mL). The combined organic phase was dried, filtered, and concentrated. The residue was purified by MPLC (SiO₂, 5 cm \times 45 cm, 3% EtOAc/hexanes) to give the monoalkylated compound **12** [5.03 g, 47%, colorless liquid, *R*_f = 0.37 (10% EtOAc/hexanes)]. Starting material recovered: 2.49 g (63% conversion). [α]_D²⁵ 147 (*c* 2.4, CHCl₃). IR (neat, cm⁻¹): 1732

($\nu_{C=O}$), 1659 ($\nu_{C=C}$). $^1\text{H NMR}$ (400 MHz, CDCl_3): δ 4.81 (s, 1H, H-2'), 4.79 (s, 1H, H-2'), 3.40 (d, 1H, $J = 3.1$ Hz, H-6), 2.49 (td, 1H, $J_1 = 11.9$ Hz, $J_2 = 4.2$ Hz, H-4), 2.21 (dt, 1H, $J_1 = 14.8$ Hz, $J_2 = 3.4$ Hz, H-5), 1.97 (2H, m, H-3 and H-5), 1.63 (3H, s, CH_3 -3'), 1.41 (3H, s, CH_3 -C-1), 1.04 (3H, d, $J = 6.9$ Hz, CH_3 -C-3). $^{13}\text{C NMR}$ (100.6 MHz, CDCl_3): 207.6 (C=O, C-2), 144.9 (=C, C-1'), 113.2 (=CH₂, C-2'), 60.6 (CH, C-6), 58.5 (C, C-1), 44.7 (CH, C-3), 42.2 (CH, C-4), 29.2 (CH₂, C-5), 18.4 (CH₃, C-3'), 15.8 (CH₃, CH_3 -C-1), 13.8 (CH₃, CH_3 -C-3). HMRS ($[\text{C}_11\text{H}_{17}\text{O}_2]^+$ ($[\text{M} + \text{H}]^+$), 181.1228; found, 181.1222.

(1R,3S,4R,6R)-4-Hydroxy-1,3-dimethyl-7-oxabicyclo[4.1.0]heptan-2-one (13a). A stream of O_3/O_2 (0.7 bar, 0.1 NL/h, 50 w) was bubbled through a -78°C cooled solution of ketone **12** (5 g, 27.74 mmol, 1 equiv) in dry MeOH (5.1 mL, 125 mmol, 4.5 equiv) and dry CH_2Cl_2 (50 mL) until the solution turned blue (30 min). The excess of O_3 was removed with a flow of argon for 30 min. The reaction mixture was allowed to reach room temperature (40 min) under a slow flow of argon and then cooled to -35°C . After 10 min, Et_3N (30 mL, 222 mmol, 8 equiv) and DMAP (0.7 g, 5.6 mmol, 0.2 equiv) were successively and slowly added. Once DMAP has dissolved, Ac_2O (21 mL, 222 mmol, 8 equiv, freshly distilled from P_2O_5 under argon) was added. The reaction mixture was allowed to reach -8°C and stirred for 2 h. The reaction was quenched by the slow addition of MeOH (20 mL). The mixture was stirred at room temperature for 5 min, diluted with EtOAc (100 mL), and successively washed with an aqueous solution of citric acid (2×50 mL, 10%) and saturated NaHCO_3 (2×50 mL). The combined organic phase was dried, filtered, and concentrated. The residue was dissolved in MeOH (60 mL). NaOAc (0.460 g, 5.55 mmol, 0.2 equiv) was added. The mixture was heated at 37°C for 12 h and then concentrated to half volume. The residue was dissolved in EtOAc (30 mL) and washed with saturated NH_4Cl (25 mL). The aqueous phase was extracted with EtOAc (2×20 mL). The combined organic phase was dried, filtered, and concentrated. The residue was purified by flash chromatography (SiO_2 , 5 cm \times 10 cm, 20% EtOAc/hexanes) to give alcohol **13a** [3.47 g, 80%, colorless oil, $R_f = 0.30$ (50% EtOAc/hexanes)]. $[\alpha]_D^{25}$ 79.8 (c 1.3, CHCl_3). IR (neat, cm^{-1}): 3445 ($\nu_{\text{O-H}}$), 1707 ($\nu_{\text{C=O}}$). $^1\text{H NMR}$ (400 MHz, CDCl_3): δ 3.79 (td, 1H, $J_1 = 13.6$ Hz, $J_2 = 9.1$ Hz, H-4), 3.38 (s, 1H, H-6), 2.62 (dt, 1H, $J_1 = 14.4$ Hz, $J_2 = 3.6$ Hz, H-5), 2.41 (s, 1H, OH), 2.01 (m, 1H, H-3), 1.92 (dd, 1H, $J_1 = 14.4$ Hz, $J_2 = 10.0$ Hz, H-5), 1.38 (s, 3H, CH_3 -C-1), 1.21 (d, 3H, $J = 7.0$ Hz, CH_3 -C-3). $^{13}\text{C NMR}$ (100.6 MHz, CDCl_3): δ 206.1 (C=O, C-2), 67.9 (CH, C-4), 59.9 (CH, C-6), 58.5 (C, C-1), 50.8 (CH, C-3), 32.1 (CH₂, C-5), 15.2 (CH₃, CH_3 -C-1), 12.5 (CH₃, CH_3 -C-3). Anal. Calcd for $[\text{C}_8\text{H}_{12}\text{O}_3]$: C, 62.06; H, 8.46. Found: C, 61.52; H, 7.94.

(1R,3S,4R,6R)-[(tert-Butyldimethylsilyloxy)-1,3-dimethyl-7-oxabicyclo[4.1.0]heptan-2-one (13b). Imidazole (2.6 g, 38.4 mmol, 2 equiv) and TBSCl (4.34 g, 28.8 mmol, 1.5 equiv) were successively added to a solution of alcohol **13a** (3 g, 19.2 mmol, 1 equiv) in dry DMF (30 mL). After 12 h, the reaction was quenched by the addition of a few pieces of ice and hexanes. The aqueous phase was extracted with hexanes (2×50 mL). The combined organic phase was dried, filtered, and concentrated. The residue was purified by flash chromatography (SiO_2 , 4 cm \times 10 cm, hexanes) to give **13b** [4.62 g, 89%, colorless oil, $R_f = 0.75$ (20% EtOAc/hexanes)]. $[\alpha]_D^{25}$ 24.2 (c 1.34, CHCl_3). IR (neat, cm^{-1}): 1712 ($\nu_{\text{C=O}}$). $^1\text{H NMR}$ (400 MHz, CDCl_3): δ 3.80 (td, 1H, $J_1 = 9.0$ Hz, $J_2 = 4.8$ Hz, H-4), 3.33 (dd, 1H, $J_1 = 3.1$ Hz, $J_2 = 1.3$ Hz, H-6), 2.49 (ddd, 1H, $J_1 = 14.6$ Hz, $J_2 = 4.6$ Hz, $J_3 = 3.1$ Hz, H-5), 2.06 (m, 1H, H-3), 1.9 (ddd, 1H, $J_1 = 14.6$ Hz, $J_2 = 9.0$ Hz, $J_3 = 1.3$ Hz, H-5), 1.38 (s, 3H, CH_3 -C-1), 1.14 (d, 3H, $J = 7$ Hz, CH_3 -C-3), 0.86 (s, 9H, $\text{Me}_3\text{C-Si}$), 0.06 (s, 3H, Me-Si), 0.04 (s, 3H, Me-Si). $^{13}\text{C NMR}$ (100.6 MHz, CDCl_3): δ 206.2 (C=O, C-2), 68.8 (CH, C-4), 60.1 (CH, C-6), 58.2 (C, C-1), 51.4 (CH, C-3), 32.8 (CH₂, C-5), 25.6 (3xCH₃, $\text{Me}_3\text{C-Si}$), 17.8 [C, (C, C-Si)], 15.3 (CH₃, CH_3 -C-1), 12.8 (CH₃, CH_3 -C-3), -4.4 (CH₃, CH_3 -Si), -4.8 (CH₃, CH_3 -Si).

HMRS ($[\text{C}_1]^+$): calculated for $[\text{C}_{14}\text{H}_{27}\text{O}_3\text{Si}]^+$ ($[\text{M} + \text{H}]^+$), 271.1729; found, 271.1724.

(1S,2S,3R,4R,6R)-4-[(tert-Butyldimethylsilyloxy)-1,3-dimethyl-7-oxabicyclo[4.1.0]heptan-2-ol (14a). A solution of L-Selectride (22 mL, 22.2 mmol, 1 M, 1.5 equiv) in dry THF was added dropwise to a solution of ketone **13b** (4 g, 14.8 mmol, 1 equiv) in dry THF (40 mL) at -78°C . After 30 min, the reaction was quenched by the dropwise addition of MeOH (10 mL) and H_2O (10 mL). A solution of NaOH (10 mL, 10%) and H_2O_2 (10 mL, 30% v/v) was successively and slowly added. The mixture was allowed to reach room temperature over 12 h. A saturated solution of NH_4Cl (30 mL) was added. The aqueous phase was extracted with a solution of EtOAc/hexanes (20%, 3×50 mL). The combined organic phase was dried, filtered, and concentrated. The residue was purified by flash chromatography (SiO_2 , 4 cm \times 10 cm, 20% EtOAc/hexanes) to give **14a** [3.51 g, 87%, colorless, $R_f = 0.4$ (20% EtOAc/hexanes)]. $[\alpha]_D^{25}$ -79.7 (c 3.7, CHCl_3). IR (neat, cm^{-1}): 3471 ($\nu_{\text{O-H}}$). $^1\text{H NMR}$ (400 MHz, CDCl_3): δ 3.82 (dd, 1H, $J_1 = 10.3$ Hz, $J_2 = 4.5$ Hz, H-2), 3.51 (td, 1H, $J_1 = 9.0$ Hz, $J_2 = 4.6$ Hz, H-4), 3.24 (sa, 1H, H-6), 2.37 (dd, 1H, $J_1 = 14.8$ Hz, $J_2 = 4.5$ Hz, H-5), 1.81 (d, 1H, $J = 10.3$ Hz, OH), 1.66 (dd, 1H, $J_1 = 14.8$ Hz, $J_2 = 9.0$ Hz, H-5), 1.49 (m, 1H, H-3), 1.42 (s, 3H, CH_3 -C-1), 0.98 (d, 3H, $J = 7.0$ Hz, CH_3 -C-3), 0.87 (s, 9H, $\text{Me}_3\text{C-Si}$), 0.04 (s, 3H, CH_3 -Si), 0.03 (s, 3H, Me-Si). $^{13}\text{C NMR}$ (100.6 MHz, CDCl_3): δ 72.8 (CH, C-2), 66.3 (CH, C-4), 63.7 (CH, C-6), 60.9 (C, C-1), 42.2 (CH, C-3), 34.9 (CH₂, C-5), 25.8 (3 \times CH₃, $\text{Me}_3\text{C-Si}$), 21.6 (CH₃, CH_3 -C-1), 18.0 (C, C-Si), 12.4 (CH₃, CH_3 -C-3), -4.4 (CH₃, Me-Si), -4.8 (CH₃, Me-Si). Anal. Calcd for $[\text{C}_8\text{H}_{12}\text{O}_3]$: C, 61.72; H, 10.36. Found: C, 61.93; H, 10.49.

(1R,2S,3S,4R,6R)-2,4-Bis[(tert-butyldimethylsilyloxy)-1,3-dimethyl-7-oxabicyclo[4.1.0]heptane (14b). Imidazole (0.63 g, 9.2 mmol, 2.5 equiv) and TBSCl (1.1 g, 7.34 mmol, 2 equiv) were successively added to a solution of alcohol **14a** (1 g, 3.67 mmol, 1 equiv) in dry DMF (10 mL). After 12 h, the reaction was quenched by the addition of ice pieces and saturated NH_4Cl (10 mL). The aqueous phase was extracted with hexanes (3×15 mL). The combined organic phase was dried, filtered, and concentrated in vacuo. The residue was purified by flash chromatography (SiO_2 , 3 cm \times 10 cm, hexanes) to give **14b** [1.37 g, colorless oil, $R_f = 0.78$ (20% EtOAc/hexanes)]. $[\alpha]_D^{25}$ -42.3 (c 4.2, CHCl_3). $^1\text{H NMR}$ (400 MHz, CDCl_3): δ 4.22 (d, 1H, $J = 5.7$ Hz, H-2), 3.79 (td, 1H, $J_1 = 9.1$ Hz, $J_2 = 4.4$ Hz, H-4), 2.97 (d, 1H, $J = 4.5$ Hz, H-6), 2.19 (dd, 1H, $J_1 = 15.5$ Hz, $J_2 = 4.1$ Hz, H-5), 1.84 (m, 1H, H-3), 1.71 (dt, 1H, $J_1 = 15.5$ Hz, $J_2 = 4.2$ Hz, H-5), 1.32 (s, 3H, CH_3 -C-1), 0.94 (s, 9H, $\text{Me}_3\text{C-Si}$), 0.93 (d, 3H, $J = 6.8$ Hz, CH_3 -C-3), 0.88 (s, 9H, $\text{Me}_3\text{C-Si}$), 0.11 (s, 3H, Me-Si), 0.07 (s, 3H, Me-Si), 0.03 (s, 6H, 2 \times Me-Si). $^{13}\text{C NMR}$ (100.6 MHz, CDCl_3): δ 71.2 (CH, C-2), 70.5 (CH, C-4), 59.8 (C, C-1), 59.5 (CH, C-6), 41.5 (CH, C-3), 31.1 (CH₂, C-5), 25.9 (3 \times CH₃, $\text{Me}_3\text{C-Si}$), 25.7 (3 \times CH₃, $\text{Me}_3\text{C-Si}$), 21.4 (CH₃, CH_3 -C-1), 18.3 (C, C-Si), 17.9 (C, C-Si), 11.7 (CH₃, CH_3 -C-3), -4.5 (CH₃, Me-Si), -4.6 (CH₃, Me-Si), -4.8 (CH₃, Me-Si), -4.9 (CH₃, Me-Si). Anal. Calcd for $[\text{C}_{20}\text{H}_{43}\text{O}_3\text{Si}_2]$: C, 62.12; H, 10.95. Found: C, 62.36; H, 11.26.

(3R,4S,5S)-3,5-Bis[(tert-butyldimethylsilyloxy)-4-methyl-6-oxoheptanal (15). H_5IO_6 (2.65 g, 11.64 mmol, 3 equiv) was added to a solution of epoxide **14b** (1.5 g, 3.73 mmol, 1 equiv) in dry Et_2O (20 mL). After 12 h, the reaction mixture was poured into an aqueous saturated solution of $\text{Na}_2\text{S}_2\text{O}_3$ (20 mL). The aqueous phase was extracted with Et_2O (3×15 mL). The combined organic phase was dried, filtered, and concentrated in vacuo to give aldehyde **11** (1.56 g), which was immediately used in the next step. $^1\text{H NMR}$ (250 MHz, CDCl_3): δ 9.71 (t, 1H, $J = 2.4$ Hz, H-1), 4.17 (c, 1H, $J = 5.4$ Hz, H-3), 3.88 (d, 1H, $J = 5.1$ Hz, H-5), 2.65 (m, 2H, CH_2 -4), 2.10 (s, 3H, CH_3 -7), 1.93 (m, 1H, H-4), 0.86 (s, 9H, $\text{Me}_3\text{C-Si}$), 0.80 (d, 3H, $J = 7.0$ Hz, CH_3 -C-4), 0.79 (s, 9H, $\text{Me}_3\text{C-Si}$), -0.01 (s, 3H, Me-Si), -0.02 (s, 3H, Me-Si), -0.03 (s, 3H, Me-Si), -0.04 (s, 3H, Me-Si). $^{13}\text{C NMR}$ (62.9 MHz, CDCl_3): δ 211.4 (C=O, C-6), 201.9 (C=O, C-1), 79.9 (CH, C-5), 69.0 (CH, C-3), 48.9 (CH₂, C-2), 44.0 (CH, C-4), 26.6 (CH₃, C-7),

25.7 (3 × CH₃, Me₃C-Si), 25.6 (3 × CH₃, Me₃C-Si), 17.9 (C, C-Si), 17.8 (C, C-Si), 12.5 (CH₃, CH₃C-4), -3.7 (CH₃, Me-Si), -4.6 (CH₃, Me-Si), -5.1, (CH₃, Me-Si), -5.2 (CH₃, Me-Si).

(3S,4S,5R)-3,5-Bis[(*tert*-butyldimethylsilyloxy)-8,8-dibromo-4-methyloct-7-en-2-one (16). CBr₄ (3.86 g, 11.64 mmol, 3 equiv) was added to a suspension of Ph₃P (3.05 g, 11.64 mmol, 3 equiv) and Zn (0.761 g, 11.64 mmol, 3 equiv) in CH₂Cl₂ (15 mL) at 0 °C. After 5 min, the cooling bath was removed and the reaction mixture was stirred at room temperature for 2 h. The color changed from green to deep-red. A solution of aldehyde **15** (1.56 g, 4.88 mmol, 1 equiv) in CH₂Cl₂ (5 mL) was added via cannula. After 45 min, the reaction mixture was filtered through a path of silica gel (elution with hexanes and 20% EtOAc/hexanes). After concentration in vacuo, the residue was purified by flash chromatography (SiO₂, 3 cm × 12 cm, 2% EtOAc/hexanes) to give the dibromide **16** [1.63 g, 75% (two steps), colorless oil, *R*_f = 0.50 (5% EtOAc/hexanes)]. [α]_D²⁵ -25.4 (*c* 2.0, CHCl₃). IR (neat, cm⁻¹): 1708 (ν_{C=O}), 1633 (ν_{C=C}). ¹H NMR (400 MHz, CDCl₃): δ 6.46 (dd, 1H, *J*₁ = 8.1 Hz, *J*₂ = 5.9 Hz, H-7), 3.88 (d, 1H, *J* = 5.6 Hz, H-3), 3.83 (c, 1H, *J* = 5.4 Hz, H-5), 2.59 (ddd, 1H, *J*₁ = 15.5 Hz, *J*₂ = 8.2 Hz, *J*₃ = 5.1 Hz, H-6), 2.3 (dt, 1H, *J*₁ = 15.5 Hz, *J*₂ = 5.7 Hz, H-6), 2.17 (s, 3H, CH₃-1), 1.81 (m, 1H, H-4), 0.94 (s, 9H, Me₃C-Si), 0.89 (d, 3H, *J* = 5.5 Hz, CH₃C-4), 0.88 (s, 9H, Me₃C-Si), 0.08 (s, 3H, Me-Si), 0.06 (s, 3H, Me-Si), 0.04 (s, 6H, 2 × Me-Si). ¹³C NMR (100.6 MHz, CDCl₃): δ 212.3 (C=O, C-2), 135.5 (CH, C-7), 90.0 (C, C-8), 79.6 (CH, C-3), 71.2 (CH, C-5), 43.6 (CH, C-4), 38.8 (CH₂, C-6), 27.0 (CH₃, C-1), 25.9 (3 × CH₃, Me₃C-Si), 25.7 (3 × CH₃, Me₃C-Si), 18.1 (C, C-Si), 18.0 (C, C-Si), 11.8 (CH₃, CH₃C-4), -4.2 (CH₃, Me-Si), -4.4 (CH₃, Me-Si), -4.8 (CH₃, Me-Si), -4.9 (CH₃, Me-Si). HMRS ([CI]⁺): calculated for [C₂₁H₄₃Br₂O₃Si]⁺, 557.1117; found, 557.1115.

(3S,4S,5R)-3,5-Bis[(*tert*-butyldimethylsilyloxy)-4-methyloct-1-en-7-yn-2-yl Trifluoromethanesulfonate (8). A freshly prepared solution of LiN(*i*-Pr)₂ in THF (6 mL, 0.9 M, 3 equiv) was added dropwise to a solution of the dibromide **16** (1 g, 1.79 mmol, 1 equiv) in THF (10 mL) at -78 °C. After being stirred for 1 h, a solution of *n*-BuLi in hexanes (0.25 mL, 0.54 mmol, 2.5 M, 0.3 equiv) was added. After 15 min, *N*-(5-chloro-2-pyridyl)bis(trifluoromethanesulfonimide) (2.1 g, 5.37 mmol, 3 equiv) was added at once. The reaction mixture was allowed to reach room temperature over 12 h. The reaction was quenched by the addition of saturated NaCl (20 mL). The aqueous phase was extracted with Et₂O (2 × 20 mL). The combined organic phase was dried, filtered, and concentrated in vacuo. The residue was purified by flash chromatography (SiO₂, 3 cm × 12 cm, hexanes) to give enoltriflate **8** [0.712 g, 75%, colorless oil, *R*_f = 0.75 (5% Et₂O/hexanes)]. [α]_D²⁵ -1.52 (*c* 2.1 mg/mL, CHCl₃). IR (neat, cm⁻¹): 3312 (ν_{=C-H}), 1663 (ν_{C=C}). ¹H NMR (500 MHz, CDCl₃): δ 5.24 (d, 1H, *J* = 3.7 Hz, H-1), 5.15 (d, 1H, *J* = 3.7 Hz, H-1), 4.10 (d, 1H, *J* = 7.3 Hz, H-3), 4.07 (ddd, 1H, *J*₁ = 7.4 Hz, *J*₂ = 5.3 Hz, *J*₃ = 3.2 Hz, H-5), 2.43 (ddd, 1H, *J*₁ = 16.8 Hz, *J*₂ = 5.3 Hz, *J*₃ = 2.7 Hz, H-6), 2.38 (ddd, 1H, *J*₁ = 1.7 Hz, *J*₂ = 7.5 Hz, *J*₃ = 2.7 Hz, H-6), 2.12 (cd, 1H, *J*₁ = 7.1 Hz, *J*₂ = 3.2 Hz, H-4), 1.99 (t, 1H, *J* = 2.7 Hz, H-8), 0.92 (s, 9H, Me₃C-Si), 0.89 (s, 9H, Me₃C-Si), 0.87 (d, 3H, *J* = 7.0 Hz, CH₃C-4), 0.11 (CH₃, Me-Si), 0.10 (CH₃, Me-Si), 0.09 (CH₃, Me-Si), 0.08 (CH₃, Me-Si). ¹³C NMR (125.7 MHz, CDCl₃): δ 156.2 (=C, C-2), 118.4 (C, *c*, *J* = 320 Hz, CF₃), 104.4 (=CH₂, C-1), 80.8 (≡CH, C-8), 74.1 (CH, C-3), 70.7 (≡C, C-7), 70.1 (CH, C-3), 42.1 (CH, C-4), 25.8 (3 × CH₃, Me₃C-Si), 25.7 (3 × CH₃, Me₃C-Si), 25.7 (CH₂, C-6), 18.2 (C, C-Si), 18.1 (C, C-Si), 9.8 (CH₃, CH₃C-4), -3.8 (CH₃, Me-Si), -4.4 (CH₃, Me-Si), -5.0 (2 × CH₃, 2 × Me-Si). HMRS ([CI]⁺): calculated for [C₂₂H₄₂F₃O₅SSi₂]⁺, 531.2244; found, 531.2239.

2α-Methyl-20S,23S-epoxymethano-1α,25-dihydroxyvitamin D₃ (4a). A solution of *t*-BuLi in pentane (0.280 mL, 0.427 mmol, 1.55 M, 2.2 equiv) was added dropwise to a solution of bromide **7a** (0.097 g, 0.194 mmol, 1 equiv) in dry THF (2 mL) at -78 °C. After the mixture was stirred for 1 h, a solution of dry ZnBr₂ in THF (0.46 mL, 0.232 mmol, 0.5 M, 1.2 equiv) was added dropwise. After 5 min, the -78 °C cooling bath was replaced

by an ice-water cooling bath and the yellow solution was stirred for 1 h. A solution of enoltriflate **8** (0.070 g, 0.132 mmol, 0.68 equiv), (Ph₃P)₄Pd (15 mg, 0.012 mmol, 6 mol %), and dry Et₃N (0.1 mL, 0.75 mmol, 4 equiv) in dry THF (2 mL) was added. The yellow color changed to orange. The reaction mixture was stirred in the dark for 5 min at 0 °C and then at room temperature for 12 h. The reaction was quenched by the addition of H₂O (1 mL). The aqueous phase was extracted with Et₂O (3 × 3 mL). The combined organic phase was dried, filtered, and concentrated in vacuo. The residue (**17a**) was dissolved in THF (2 mL) under argon. A solution of *n*-Bu₄NF in THF (1 mL, 0.97 mmol, 1 M, 5 equiv) was added. The mixture was stirred for 24 h. The reaction was quenched by the addition of saturated NH₄Cl (5 mL). The aqueous phase was extracted with EtOAc (3 × 10 mL). The combined organic phase was dried, filtered, and concentrated in vacuo. The residue was purified by flash chromatography (SiO₂, 1.5 cm × 10 cm, 50% EtOAc/hexanes) to give **4a** [0.033 g, 55%, white solid, *R*_f = 0.48 (60% EtOAc/hexanes)]. ¹H NMR (500 MHz, CDCl₃): δ 6.37 (d, 1H, *J* = 11.3 Hz, H-6), 6.01 (d, 1H, *J* = 11.3 Hz, H-7), 5.27 (s, 1H, H-19), 5.00 (d, 1H, *J* = 1.6 Hz, H-19), 4.31 (d, 1H, *J* = 3.1 Hz, H-1), 4.10 (t, 1H, *J* = 7.9 Hz, H-28), 3.84 (td, 1H, *J*₁ = 7.5 Hz, *J*₂ = 4.3 Hz, H-3), 3.34 (dd, 1H, *J*₁ = 9.7 Hz, *J*₂ = 8.5 Hz, H-28), 2.81 (d, 1H, *J*₁ = 12.5 Hz, H-9), 2.66 (dd, 1H, *J*₁ = 13.5 Hz, *J*₂ = 3.8 Hz, H-4), 2.46 (m, 1H, H-23), 2.22 (dd, 1H, *J*₁ = 13.5 Hz, *J*₂ = 7.6 Hz, H-4), 2.05 (d, 1H, *J* = 11.7 Hz, H-12), 1.99 (dd, 1H, *J*₁ = 11.2 Hz, *J*₂ = 7.9 Hz, H-14), 1.92 (cd, 1H, *J*₁ = 7.0 Hz, *J*₂ = 3.7 Hz, H-2), 1.23 (s, 3H, CH₃-21), 1.22 (s, 6H, CH₃-26 and CH₃-27), 1.06 (d, 3H, *J* = 6.9 Hz, CH₃C-2α), 0.66 (s, 3H, CH₃-18). ¹³C NMR (125.7 MHz, CDCl₃): δ 146.5 (=C, C-10), 142.7 (=C, C-8), 133.1 (=C, C-5), 124.7 (=CH, C-6), 117.3 (=CH, C-7), 113.1 (=CH₂, C-19), 84.9 (C, C-20), 75.2 (CH, C-1), 73.2 (CH₂, C-28), 71.7 (CH, C-3), 70.9 (CH, C-25), 59.8 (CH, C-14), 56.5 (CH, C-17), 47.1 (CH₂, C-22), 46.9 (CH₂, C-24), 46.1 (C, C-13), 44.1 (CH₂, C-2), 43.2 (CH₂, C-4), 40.7 (CH₂, C-12), 35.9 (CH, C-23), 30.2 (CH₃, C-27), 29.7 (CH₃, C-26), 29.0 (CH₂, C-9), 27.6 (CH₃, CH₃-21), 23.4 (CH₂), 22.7 (CH₂), 21.9 (CH₂), 13.2 (CH₃, C-18), 12.4 (CH₃, CH₃C-2α). HMRS (ESI-TOF): calculated for [C₂₉H₄₇O₄]⁺ ([M + H]⁺), 459.3475; found, 459.3469.

2α-Methyl-20S,23R-epoxymethano-1α,25-dihydroxyvitamin D₃ (4b). A solution of *t*-BuLi in pentane (0.370 mL, 0.630 mmol, 1.7 M, 2.1 equiv) was added dropwise to a solution of bromide **7b** (0.150 g, 0.3 mmol, 1 equiv) in dry THF (2 mL) at -78 °C. After the mixture was stirred for 1 h, a solution of dry ZnBr₂ in THF (0.5 mL, 0.41 mmol, 0.82 M, 1.37 equiv) was added dropwise. After 5 min, the -78 °C cooling bath was replaced by an ice-water cooling bath and the yellow solution was stirred for 1 h. A solution of enoltriflate **8** (0.080 g, 0.150 mmol, 0.5 equiv), (Ph₃P)₄Pd (20 mg, 0.017 mmol, 6 mol %), and dry Et₃N (0.2 mL, 1.5 mmol, 5 equiv) in dry THF (2 mL) was added. The yellow color changed to orange. The reaction mixture was stirred in the dark for 5 min at 0 °C and then at room temperature for 6 h. The reaction was quenched by the addition of H₂O (1 mL). The aqueous phase was extracted with Et₂O (3 × 3 mL). The combined organic phase was dried, filtered, and concentrated in vacuo. The residue (**17b**) was dissolved in a mixture of deoxygenated CH₂Cl₂ (2 mL), CH₃CN (2 mL), and Et₃N (1 mL). HF-Py complex (0.2 mL) was added. The mixture was stirred at room temperature for 2 h. The reaction was quenched by the addition of saturated NaHCO₃ (5 mL). The aqueous phase was extracted with EtOAc (3 × 10 mL). The combined organic phase was dried, filtered, and concentrated in vacuo. The residue was purified by flash chromatography (SiO₂, 1.5 cm × 10 cm, 50% EtOAc/hexanes) and then by HPLC (Phenomenex Silica(2), 250 mm × 1.2 mm, 10% *i*-PrOH/hexanes) to give **4b** [0.036 g, 53%, white solid, *R*_f = 0.48 (60% EtOAc/hexanes)]. ¹H NMR (500 MHz, CDCl₃): δ 6.37 (d, 1H, *J* = 11.3 Hz, H-6), 6.00 (d, 1H, *J* = 11.3 Hz, H-7), 5.27 (s, 1H, H-19), 5.00 (d, 1H, *J* = 1.9 Hz, H-19), 4.31 (d, 1H, *J* = 3.6 Hz,

H-1), 4.07 (t, 1H, $J = 7.7$ Hz, H-28), 3.84 (td, 1H, $J_1 = 7.5$ Hz, $J_2 = 4.2$ Hz, H-3), 3.37 (dd, 1H, $J_1 = 9.8$ Hz, $J_2 = 8.9$ Hz, H-28), 2.81 (dd, 1H, $J_1 = 11.9$ Hz, $J_2 = 3.7$ Hz, H-9), 2.66 (dd, 1H, $J_1 = 13.6$ Hz, $J_2 = 3.9$ Hz, H-4), 2.32 (m, 1H, H-23), 1.27 (s, 3H, CH₃-21), 1.21 (s, 6H, CH₃-26 and CH₃-27), 1.06 (d, 3H, $J = 6.9$ Hz, CH₃-2), 0.66 (s, 3H, CH₃-18). ¹³C NMR (125.7 MHz, CDCl₃): δ 146.5 (=C, C-10), 142.7 (=C, C-8), 133.1 (=C, C-5), 124.7 (=CH, C-6), 117.3 (=CH, C-7), 113.1 (=CH₂, C-19), 84.9 (C, C-20), 75.2 (CH, C-1), 73.2 (CH₂, C-28), 71.7 (CH, C-3), 70.9 (CH, C-25), 59.8 (CH, C-14), 56.5 (CH, C-17), 47.1 (CH₂, C-22), 46.9 (CH₂, C-24), 46.1 (C, C-13), 44.1 (CH₂, C-2), 43.2 (CH₂, C-4), 40.7 (CH₂, C-12), 35.9 (CH, C-23), 30.2 (CH₃, C-27), 29.7 (CH₃, C-26), 29.0 (CH₂, C-9), 27.6 (CH₃, CH₃-21), 23.4 (CH₂), 22.7 (CH₂), 21.9 (CH₂), 13.2 (CH₃, C-18), 12.4 (CH₃, CH₃-2 α). HMRS (ESI-TOF): calculated for [C₂₉H₄₇O₄]⁺ ([M + H]⁺), 459.3475; found, 459.3478.

Purification and Crystallization. Purification and crystallization of the human VDR LBD complexes with **4a** or **4b** were performed as previously described.⁶ The LBD of the human VDR (residues 118–427, Δ 166–216) was cloned in pET28b expression vector to obtain an N-terminal hexahistidine-tagged fusion protein and was overproduced in *E. coli* BL21 (DE3) strain. Cells were grown in Luria–Bertani medium and subsequently incubated for 3 h at 25 °C with 1 mM isopropyl thio- β -D-galactoside. The protein purification included a metal affinity chromatography step on a cobalt-chelating resin (TALON, Clontech). The tag was removed by thrombin digestion overnight at 4 °C, and the protein was further purified by gel filtration on a Superdex S200 16/60 column (Amersham). The protein buffer prior to concentration of the protein contains 20 mM Tris, pH 7.5, 200 mM NaCl, and 2 nM TCEP. The protein was concentrated to 3.5 mg/mL and incubated in the presence of a 1.5-fold excess of ligands. The purity and homogeneity of the protein were assessed by SDS–PAGE. Crystals of complexes were obtained at 4 °C by vapor diffusion method using crystals of VDR LBD/1 α ,25(OH)₂D₃ as microseeds. The reservoir solution contained 0.1 M Mes and 1.4 M ammonium sulfate at pH 6.0.

X-ray Data Collection and Structure Determination. The crystals were mounted in fiber loop and flash-cooled in liquid nitrogen after cryoprotection with a solution containing the reservoir solution plus 30% glycerol and 2% polyethylene glycol 400. Data collection from a single frozen crystal was performed at 100 K on beamlines ID23-2 and ID29 of the ESRF (Grenoble, France) for **4a** and **4b** complexes, respectively. The crystals belong to the orthorhombic space group *P*2₁2₁2₁ with one monomer per asymmetric unit. Data were integrated and scaled using HKL2000.³⁵ A rigid body refinement was used with the structure of the hVDR LBD/1 α ,25(OH)₂D₃ complex as a starting model. Refinement involved iterative cycles of manual building and refinement calculations. The programs Refmac³⁶ and COOT³⁷ were used throughout structure determination and refinement. The ligand molecule was included only at the last stage of the refinement. The omit map from the refined atomic model of VDR LBD was used to fit the ligand to its electron density, shown in Figure 2A. Anisotropic scaling and a bulk solvent correction were used. Individual *B* atomic factors were refined isotropically for the **4a** complex and anisotropically for the **4b** complex. Solvent molecules were then placed according to unassigned peaks in the difference Fourier map. In the VDR/**4a** complex, refined at 1.7 Å with no σ cutoff, the final model consists of residues 118–423 (Δ 166–216), the ligand, two sulfate ions, and 447 water molecules. In the VDR/**4b** complex refined at 1.45 Å with no σ cutoff, the final model consists of residues 118–423 (Δ 166–216), the ligand, two sulfate ions, and 458 water molecules. According to PROCHECK,³⁸ 93.0% of peptide lies in the most favored regions and 7.0% in additional allowed regions for the VDR/**4a** complex, and 93.4% of peptide lies in most favored regions and 6.6% in additional allowed regions for the VDR/**4b** complex. Data are summarized in

Supporting Information Table 1. The volumes of each ligand binding pocket and each ligand were calculated as previously reported.

Steady-State Fluorescence Anisotropy Measurements. Steady-state anisotropy measurements were performed with a T-format SLM 8000 spectrofluorometer. Anisotropy titrations were carried out by adding increasing hVDR LBD concentrations to 1 μ M fluorescent labeled TAMRA (tetramethylrhodamine)-SRC-1 (RHKILHRLLLQEGSPS) peptide in 20 mM Tris-HCl (pH 7.5), 200 mM NaCl. The binding affinity constants of the SRC-1 peptide to hVDR LBD was determined in response to 1 α ,25(OH)₂D₃, **2a**, **4a**, or **4b**. The excitation wavelength was 550 nm, and the emitted light was monitored through high-pass filters (550 nm) (Kodak). A home-built device ensured the automatic rotation of the excitation polarizer. Assuming that hVDR binds with the TAMRA-SRC-1 peptide in a 1:1 stoichiometry, the model used to describe the binding experiment was the following:



L and R represent the hVDR LBD and the TAMRA-SRC-1 peptide, respectively, and P designates the hVDR LBD/TAMRA-SRC-1 complex.

The Scatchard equation was rewritten to fit the anisotropy, *r*, as follows:

$$r = r_0 + (r_f - r_0) \left(\frac{(R_t + L_t + K_d) - \sqrt{(R_t + L_t + K_d)^2 - 4R_t L_t}}{2R_t} \right)$$

L_t and R_t are the total concentrations of hVDR LBD and TAMRA-SRC-1 peptides, respectively; *r_f* represents the anisotropy at the plateau when all the complex is formed, whereas *r₀* and *r* correspond to the anisotropy values of the TAMRA-SRC-1 peptide in the absence and in the presence of a given concentration of hVDR, respectively. *K_d* corresponds to the dissociation constant of the complex. All experiments were performed at 20 °C, and results are representative of three distinct experiments.

Transient Transfections and Luciferase Reporter Gene Assays. The chimera Gal4-VDR LBD (90–427) was constructed by PCR using the appropriate oligonucleotides with restriction sites (XhoI/BamHI) and the pSG5 VDR plasmid (1–427) as template and cloned into the vector PXJ440 encoding the DBD of the yeast activator Gal4 (1–147). MCF-7 human breast cancer cells were seeded into 24-well plates (10⁵ cells per well) and grown overnight in phenol red-free Dulbecco's modified Eagle's medium (DMEM) supplemented with 10% charcoal-treated fetal bovine serum (FCS), 5% gentamycin, and 0.6 μ g/mL insulin. Liposomes were formed using the transfection reagent jetPEI (Polyplus transfection) and used according to the manufacturer's instructions. Cells were transfected with 25 ng of the expression plasmid pXJ440-Gal4DBD-hVDR (90–427), 250 ng of the reporter plasmid ptk-LUC, 50 ng of the pCH110- β -galactosidase vector (used as an internal control to normalize variation in transfection efficiency), and 675 ng of the carrier plasmid pBluescript (Stratagene). Ten hours later, cells were washed with freshly prepared phosphate buffered saline (PBS) and various concentrations of 1 α ,25(OH)₂D₃, **4a**, **4b**, **2a**, and **2b** versus control (solvent) were added to the cells in phenol red-free DMEM supplemented with 10% FCS. Twenty-four hours after the onset of stimulation, cells were rinsed in PBS and lysed in 100 μ L of reporter gene lysis buffer (Roche Diagnostics). Cell extracts were assayed for luciferase and β -galactosidase activities. Luciferase reporter activity was determined by using luciferin as the substrate, and β -galactosidase activity was measured using *o*-nitrophenyl- β -galactoside as the substrate. The chemiluminescence from activated luciferin was measured on a luminometer plate reader LB96P (Berthold

Technologies). Luciferase values were normalized to the β -galactosidase activity. Luciferase activities are expressed as arbitrary units of light intensity. Data points represent the mean of assays performed in triplicate for at least three independent experiments. For every triplicate, the mean and the standard deviation of the mean were calculated.

HL60 Cell Culture and Differentiation. HL60 cells were cultured in complete RPMI with 10% fetal calf serum (Sigma) and 5% gentamycin (Kalys) at 37 °C without CO₂ as previously described.¹⁹ Cells were plated at a density of 4×10^5 cells/mL and cultured for 96 h in the presence of 1 α ,25(OH)₂D₃, **2a**, or **4a** and **4b** at various concentrations. Control incubations were performed with 0.7% ethanol. HL60 cell differentiation was determined by flow cytometry using PE-conjugated antihuman CD11c and FITC-conjugated antihuman CD14 antibodies (Pharmingen/BD Biosciences). Topro-3 (Molecular probes) was added immediately before laser excitation to exclude dead cells. Cells were analyzed on a FACSCalibur (Becton Dickinson) and using the FlowJo (TreeStar Inc.) software.

Mice and Serum Calcium Quantitation. Animal protocols were approved by the Alsace Regional Ethics Committee. Male C57BL/6J mice were obtained from Charles River Laboratories, France (L'Arbresle, France). Mice (6–7 weeks old) were maintained in a temperature-controlled (23 °C) facility with a 12 h light/dark cycle and were given free access to food and water. Mice were fed with the standard mouse chow (D03 SAFE; Augy, France) and tap water. The different 1 α ,25(OH)₂D₃ agonists were dissolved in sesame oil and administered intraperitoneally every 2 days at a dose of 0.1 μ g/kg. Mice were fasted 3 h prior to blood harvest, and subsequent calcium measurements were determined with a kit supplied by Olympus according to the manufacturer's procedure on an Olympus AU400 analyzer (Olympus SA, Ringis, France).³⁹

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Supporting Information Available: Table 1 listing crystallographic data and refinement statistics, experimental procedures for the chemical synthesis, and NMR spectra of compounds. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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